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(54) Primer / primer surfacer

(57) The present invention provides for a primer composition that will lower the VOC, maintain excellent performance of the primer, and improve the mix ratio. The primer composition contains a mixture of methyl

ethyl ketone peroxide (MEKP) with a solvent. The solvent keeps the MEKP from decomposing while the potency to cure the prime is kept in the same required time.

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BACKGROUND OF THE INVENTION

[0001] The present invention relates to primers/primer surfacers, that will be applied to any substrate that will be painted and more particularly to metal, fiberglass, SMC, plastic or wood that will be painted, and it particularly relates to a primer/primer surfacer to be used in the field of automotive, marine, agricultural and industrial equipment.

[0002] There are many primer/primer surfacers on the market which used as primer to prepare the surface to be painted. Traditional primers are being gradually pushed out of the market due to changing environmental regulations. As a result, trying to create a new, volatile organic compound (VOC) compliant, high performance and user friendly primer has become important for manufacturers, such as manufacturers of automotive, marine, agricultural and industrial equipment.

[0003] Traditional urethane technology has thus far not been successfully used because of problems integrating high solids, and exempt (VOC compliant) solvent into urethane primers. Shrinkage and other poor performance characteristics have not been overcome. The epoxy system also suffers from poor performance characteristics, including long dry time and poor sanding. Acrylics (air-dry) systems have inherent VOC and shrinkage problem, and poor integrity.

[0004] The traditional polyester system brought solid performance at VOC complaint levels, with most of the benefits of epoxy and urethanes obtained without the liabilities of shrink, slow dry/cure, and poor sanding. The problem with traditional polyester primer technology is the catalyzation amount. The catalyzation amount was very subjective and prone to error. In addition, the ketone peroxide, such as methyl ethyl ketone-peroxide (MEKP), traditionally used as the catalyst has oxidative and corrosive properties that present some risks from a health and safety point of view. Also, the concentrated peroxide is susceptible to premature decomposition, gassing, and loss of reactive properties (active oxygen) if stored improperly.

[0005] There have been attempts to overcome these deficiencies by using vinyl polyester (VPE). Vinyl ester, polyester, and VPE are cured with MEKP at 2% by weight. One problem is to find a way to stabilize the MEKP so that it can be mixed with the primer in a ratio other than 2%.

[0006] Thus, there is a need in the art for a new polyester system that will lower the VOC, maintain the excellent performance of a primer, and improve the mix ratio.

SUMMARY OF THE INVENTION

[0007] This invention fulfills that need by providing a primer composition that will tower the VOC, maintain ex-

cellent performance of the primer, and improve the mix ratio. This is accomplished by mixing methyl ethyl ketone peroxide (MEKP) with a solvent which is either ethyl acetate, methyl acetate, t-butyl acetate, or mixtures thereof. The solvent keeps the MEKP from decomposing while keeping the potency to cure the primer in the same required time.

[0008] In accordance with one embodiment of the present invention, a primer composition is provided comprising an ester, methyl ethyl ketone peroxide, and a sufficient amount of solvent to stabilize the methyl ethyl ketone peroxide. The ester in primer composition is selected from the groups consisting of polyester, vinyl ester, vinyl polyester, and mixtures thereof. The solvent in the primer composition is mixed with the methyl ethyl ketone peroxide.

[0009] In accordance with another embodiment of the present invention, a method of stabilizing a primer composition is provided comprising the steps of mixing methyl ethyl ketone peroxide with a sufficient amount of solvent to stabilize the methyl ethyl ketone peroxide, then adding the mixture to a primer composition. The solvent used in the method of stabilizing a methyl ethyl ketone peroxide catalyst is selected from the group consisting of ethyl acetate, methyl acetate, t-butyl acetate, and mixtures thereof.

[0010] In accordance with another embodiment of the present invention, a stabilized methyl ethyl ketone peroxide catalyst is provided comprising methyl ethyl ketone peroxide and a sufficient amount of a solvent to stabilize the methyl ethyl ketone peroxide. The solvent in the stabilized methyl ethyl ketone peroxide catalyst is selected from the group consisting of ethyl acetate, methyl acetate, t-butyl acetate, and mixtures thereof.

DETAILED DESCRIPTION OF THE PREFERRED EMBODIMENTS

[0011] The present invention utilizes a stabilized ME-KP in order to create a new polyester system that lowers the VOC, maintains the excellent performance of a primer, and improves the mix ratio. The stabilized MEKP is produced by adding a solvent to the MEKP. More specifically, by adding either ethyl acetate, methyl acetate, t-butyl acetate, or mixtures thereof to MEKP, the MEKP is more stable and provides a longer shelf life.

[0012] The primer composition comprises an ester, a methyl ethyl ketone peroxide, and a sufficient amount of solvent to stabilize the methyl ethyl ketone peroxide. The ester can be polyester, vinyl ester, vinyl polyester, and mixtures thereof. The amount of the ester is in the range of 20-40%. The amount of methyl ethyl ketone peroxide used is in the range of 1-5%.

[0013] The solvent in the primer composition is mixed with the methyl ethyl ketone peroxide to form a solution. The solution is then added to the primer, giving the primer a shelf life over 6 months. The solvent is selected from ethyl acetate, methyl acetate, t-butyl acetate, and

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mixtures thereof.

[0014] The method of stabilizing the primer composition comprises the step of mixing methyl ethyl ketone peroxide with a sufficient amount of solvent to stabilize the methyl ethyl ketone peroxide, then adding the mixture to the primer composition.

[0015] The stabilized methyl ethyl ketone peroxide catalyst comprises methyl ethyl ketone peroxide and a sufficient amount of a solvent to stabilize the methyl ethyl ketone peroxide. The amount of methyl ethyl ketone peroxide catalyst used is in the range of 15-40%. The shelf life of the stabilized methyl ethyl ketone peroxide catalyst is over 6 months.

Claims

1. A primer composition comprising:

an ester; methyl ethyl ketone peroxide; and a sufficient amount of solvent to stabilize the methyl ethyl ketone peroxide.

- A primer composition as claimed in claim 1 wherein said ester is selected from the group consisting of polyester, vinyl ester, vinyl polyester, and mixtures thereof.
- A primer composition as claimed in claim 1 or 2 wherein the amount of said ester is between 20% and 40%.
- A primer composition as claimed in at least one of the preceding claims wherein the amount of said methyl ethyl ketone peroxide is between 1% and 5%.
- A primer composition as claimed in at least one of the preceding claims wherein said solvent is mixed with said methyl ethyl ketone peroxide to form a solution.
- 6. A primer composition as claimed in at least one of the preceding claims wherein said solvent is selected from the group consisting of ethyl acetate, methyl acetate, t-butyl acetate, and mixtures thereof.
- A primer composition as claimed in at least one of the preceding claims wherein the shelf life of said primer is over 6 months.
- 8. A method of stabilizing a methyl ethyl ketone peroxide catalyst for a primer composition comprising the steps:

mixing methyl ethyl ketone peroxide with a sufficient amount of solvent to stabilize the methyl

ethyl ketone peroxide prior to adding the mixture to a primer composition.

- A method of stabilizing a methyl ethyl ketone peroxide catalyst as claimed in claim 8 wherein said solvent is selected from the group consisting of ethyl acetate, methyl acetate, t-butyl acetate, and mixtures thereof.
- 10. A method of stabilizing a methyl ethyl ketone peroxide catalyst as claimed in claim 8 or 9 wherein the amount of methyl ethyl ketone peroxide is between 5% and 20%.
- 15 11. A method of stabilizing a methyl ethyl ketone peroxide catalyst as claimed in at least one of claims 8 to 10 wherein the shelf life of said methyl ethyl ketone peroxide catalyst is over 6 months.
- 20 12. A stabilized methyl ethyl ketone peroxide catalyst comprising:

methyl ethyl ketone peroxide; and a sufficient amount of a solvent to stabilize the methyl ethyl ketone peroxide.

- 13. A stabilized methyl ethyl ketone peroxide catalyst as claimed in claim 12 wherein said solvent is selected from the group consisting of ethyl acetate, methyl acetate, t-butyl acetate, and mixtures thereof.
- 14. A stabilized methyl ethyl ketone peroxide catalyst as claimed in claim 12 or 13 wherein the amount of methyl ethyl ketone peroxide catalyst is between 15% and 40%.
- 15. A stabilized methyl ethyl ketone peroxide catalyst as claimed in at least one of claims 12 to 14 wherein the shelf life of said stabilized methyl ethyl ketone peroxide catalyst is over 6 months.

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EUROPEAN SEARCH REPORT

Application Number EP 02 00 1569

Category	Citation of document with in of relevant pass	ndication, where appropriate, ages	Relevant to daim	CLASSIFICATION OF THE APPLICATION (Int.C).7)
X	GB 927 786 A (HOWAR 6 June 1963 (1963-0 * page 2, paragraph * page 2, line 77-8	6-06) 1 *	1-4,6,8	C09D5/00 C09D167/06 C07C407/00
X	GB 786 406 A (HUELS 20 November 1957 (1 * Example *	CHEMISCHE WERKE AG) 957-11-20)	1-4,6,8	
X ,P	US 6 265 027 B1 (J0 ET AL) 24 July 2001 * column 3, line 11		1-4,6	
Α	DATABASE WPI Section Ch, Derwent Publication Class E17, AN 1970- XP002202526 & JP 45 003367 B (N LTD) * abstract *		8,9	TECHNICAL FIELDS SEARCHED (Int.Cl.7)
A	DATABASE WPI Section Ch, Week 19 Derwent Publication Class A60, AN 1974- XP002202527 & NL 7 401 405 A (A 25 April 1974 (1974 * abstract *	s Ltd., London, GB; 35814V KZO NV),	8,9	C09D C07C C08F
	The present search report has			
	Place of search	Date of completion of the search		Examiner
	THE HAGUE	18 June 2002	Gir	ard, Y
X : part Y : part doct A : tech O : non	ATEGORY OF CITED DOCUMENTS icularly relevant if taken alone icularly relevant if combined with anot ument of the same category inological background written disclosure mediate document	E : eaflier patent after the filing her D : document cite L : document cite	d in the application d for other reasons	ished on, or

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ANNEX TO THE EUROPEAN SEARCH REPORT ON EUROPEAN PATENT APPLICATION NO.

EP 02 00 1569

This annex lists the patent family members relating to the patent documents cited in the above-mentioned European search report. The members are as contained in the European Patent Office EDP file on The European Patent Office is in no way liable for these particulars which are merely given for the purpose of information.

18-06-2002

	Patent document cited in search report		Publication date		Patent family member(s)	Publication date
	927786	Α	06-06-1963	NONE		
GB	786406	Α	20-11-1957	NONE		
US	6265027	B1	24-07-2001	NONE		
	45003367	В		NONE		
NL	7401405	Α	25-04-1974	NONE		

For more details about this annex : see Official Journal of the European Patent Office, No. 12/82

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